

Metabolomics Reveal the “Invisible” Responses of Spinach Plants Exposed to CeO₂ Nanoparticles

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3 **Spinach Plants Exposed to CeO₂ Nanoparticles**

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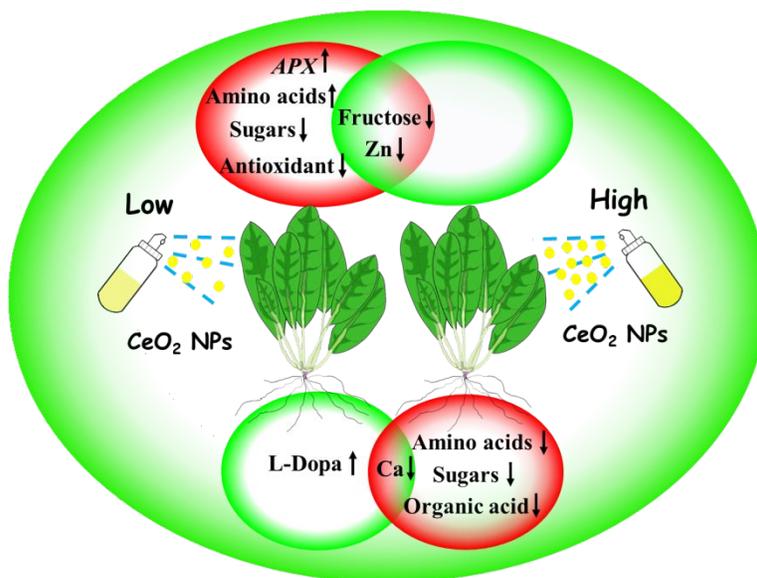
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35 **ABSTRACT**

36 Engineered nanoparticle (NP) with activities that mimic antioxidant enzymes have good
37 prospects in agriculture because they can increase photosynthesis and improve stress
38 tolerance. Here, the interaction between cerium oxide NPs with spinach plants (*Spinacia*
39 *oleracea*) was investigated by integrating phenotypic and metabolomic analyses. Soil
40 grown four-week-old spinach plants were foliar exposed for three weeks to CeO₂ NPs at
41 0, 0.3, and 3 mg per plant. Phenotypic parameters (chlorophyll fluorescence,
42 photosynthetic pigment contents, plant biomass, lipid peroxidation and membrane
43 permeability) were not affected. However, metabolomics analysis revealed that both
44 doses of CeO₂ NPs induced metabolic reprogramming in leaves and roots in a non-dose-
45 dependent manner. The low dose of CeO₂ NPs (0.3 mg per plant) induced stronger
46 metabolic reprogramming in spinach leaves than high dose of CeO₂ NPs. However, the
47 high dose of CeO₂ NPs triggered more metabolic changes in roots, compared to the low
48 dose. Foliar spray of CeO₂ NPs at 3 mg/plant induced marked down-regulation of a
49 number of amino acids (threonine, tryptophan, L-cysteine, methionine, cycloleucine,
50 aspartic acid, asparagine, tyrosine, and glutamic acid). In addition, Zn decreased by 44%
51 and 54% in leaves and Ca decreased by 38% and 32% in roots under exposure to CeO₂
52 NPs at 0.3 and 3 mg/plant, respectively. These results provide better understanding of the
53 intrinsic phenotypic and metabolic changes imposed by CeO₂ NPs in spinach plants.

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58 INTRODUCTION

59 Nanotechnology provides tools to improve plant production and protection, enabling a
60 more sustainable agriculture.¹ Nano-enabled agrochemicals have been designed to
61 encapsulate active ingredients within polymers enabling slow and target controlled
62 release.² In addition, the unique physiochemical and photoelectric properties of
63 engineered nanoparticles (NPs) can be used to enhance plant photosynthesis and growth.³
64 ⁴ For example, single-walled carbon nanotubes have been reported to increase plant
65 photosynthesis via enhanced energy capture.³ CeO₂ NPs, a lanthanide metal oxide,
66 exhibit excellent reactive oxygen species (ROS) scavenging activity.⁵⁻⁷ The ROS
67 catalytic scavenging properties of CeO₂ NPs have been utilized to protect plant
68 photosynthesis from oxidative stress and improve plant growth performance, especially
69 under stress condition.⁸

70 Recently, CeO₂ NPs (10 nm, -17 mV, Ce³⁺/Ce⁴⁺ ratio of 35%, 50 mg/L) were found
71 to enhance leaf mesophyll K⁺ retention, therefore improving *Arabidopsis thaliana*
72 salinity tolerance.⁹ In addition to improve plant performance under stress conditions, both
73 bare (10 nm, +45 mV, 8% Ce³⁺ ions) and polyvinylpyrrolidone-coated CeO₂ NPs (19 nm,
74 -52 mV, 12% of Ce³⁺ ions), at 100 mg/kg, have been reported to increase photosynthesis
75 in soybean plants.¹⁰ Furthermore, CeO₂ NPs have been applied to combat pathogens.
76 Adisa et al.¹¹ reported that CeO₂ NPs (8 nm) at 250 mg/L significantly suppressed fungal
77 pathogen (*Fusarium Wilt*) in tomato (*Solanum lycopersicum*). In contrast, CeO₂ NPs (20
78 nm, +43 mV in DI water) have been reported to induce oxidative stress, lipid
79 peroxidation and membrane damage in *Arabidopsis thaliana*¹² and *Lactuca sativa*.¹³
80 Elevated antioxidative capacity also affects plant metabolism processes. Thus far, the

81 mechanism by which the CeO₂ NPs modulate plant metabolism is not understood. Rico et
82 al.¹⁴ found changes in the physiology and nutrient profile of second generation wheat
83 plants, when the first generation plants were exposed to 500 mg CeO₂ NPs/kg. The
84 reason for such changes remains unclear. The understanding of the mechanism of
85 interaction between CeO₂ and physiological-biochemical functions of the plants is
86 imperative for their sustainable application in agriculture.^{15, 12, 16}

87 Much of the existing plant-NP interaction studies use relatively insensitive endpoints
88 (germination, biomass and pigment content), which offer limited guidance to understand
89 the interaction mechanisms between plants and NPs.¹⁷ In contrast, “omic” based
90 endpoints, such as transcriptomics, proteomics and metabolomics, can provide highly
91 detailed, global and mechanistic information on plant responses to NP.¹⁷ To date, only
92 Reichman et al.¹⁸ reported transcriptome profile changes of plants (Douglas-Fir) in
93 response to CeO₂ NPs. Metabolomics allows the identification and quantification of
94 thousands of small molecules in unexposed and NP exposed plants. Unlike other omics,
95 low molecular weight metabolites are the closest link to phenotype. In addition,
96 metabolites pool in organisms is much smaller than gene and protein pool. This
97 technology can be used as a powerful tool to understand the molecular response of plants
98 that grow under NP exposure. By using gas chromatography-mass spectrometry (GC-MS)-
99 based metabolomics, we contributed to clarify the toxicity and detoxification mechanisms
100 of CuNPs and AgNPs to *Cucumis sativus* plants.^{19, 20} Therefore, using more sensitive
101 endpoints can get deeper insights into the mechanism of plant responses to CeO₂ NPs,
102 which is important for their sustainable application and safety design.

103 In this study, laboratory synthesized CeO₂ NPs (~4 nm, 20 mV in 10 mg/L DI water)
104 were exposed for three weeks to four-week-old spinach plants at 0, 0.3, and 3 mg/plant.
105 The physiological and biochemical parameters, including photosynthesis, biomass, and
106 lipid peroxidation, were determined. GC-MS was applied to determine hundreds of low
107 molecular metabolites in leaves and roots. The phenotypic analysis, combined with
108 metabolic analysis, enabled a deep insight into the interaction between CeO₂ NPs and
109 spinach plants. The obtained results are valuable for using CeO₂ nanomaterials in a safe
110 and sustainable manner in agriculture.

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112 MATERIALS AND METHODS

113 **CeO₂ Nanoparticles Synthesis and Characterization.** The CeO₂ NPs were synthesized
114 as reported previously by Sreeremya et al.²¹ More details regarding the synthesis method
115 are provided in the Supporting Information. The synthesized CeO₂ NPs were
116 characterized by transmission electron microscope (TEM) (JEM-200CX, JEOL, Japan)
117 for for size, shape and morphology. The hydrodynamic diameter and zeta potential of
118 CeO₂ NPs in exposure medium (10 and 100 mg/L in nanopure water) were measured
119 using a Nano Zetasizer (Malvern, Nanosizer ZS90, USA). The CAT-mimic activity of
120 CeO₂ NPs was determined according to Wang et al.²² Specifically, 20 μL CeO₂ NPs stock
121 solution (50 mg/L and 500 mg/L, final concentration is 10 mg/L and 100 mg/L) was
122 mixed with 80 μL H₂O₂ solution (100 mg/L) and incubated for 3 h at 37 °C. 50 μL
123 sulfuric acid (9 mol/L) was added subsequently to stop the reaction. Then 50 μL KMnO₄
124 (3.2 mmol/L) was added to react with remaining H₂O₂. The absorbance at 525 nm was
125 determined using a microplate spectrophotometer (Varioskan LUX, Thermo scientific,

126 America). This experiment was repeated in triplicate. The superoxide dismutase (SOD)
127 mimic activities of CeO₂ NPs were determined using SOD assay kit (Nanjing Jiancheng
128 Bioengineer Ins., Nanjing, China), following the manufacturer's instructions. More
129 details regarding the SOD assays are provided in Supporting Information.

130 **Plant Growth and CeO₂ Nanoparticles Exposure.** Spinach (*Spinacia oleracea*) seeds
131 were purchased from Hezhiyuan Seed Inc. (Shandong, China). The spinach accession
132 Octavia was provided by Syngenta and their F1 hybrid were used for this study. The
133 organic potting soil Scotts Miracle-Gro (Scotts Co, Marysville, Ohio) as growth medium.
134 The nutrient element content in potting soil was shown in **Table S1**. Plastic pot
135 containing 100 g of soil were planted with one/pot. The plants were cultivated in a green
136 house for 28 days at 25 °C during the day and 20 °C at night. The daily light integral was
137 180 μmol·m⁻²·s⁻¹. Since the potting soil is rich of nutrients, no additional nutrients were
138 added during whole growth period. Four weeks old spinach plants were foliar applied
139 with CeO₂ NPs (10 and 100 mg/L, 0.3 and 3 mg/plant as total amount applied) twice per
140 week for three weeks. For each foliar application, CeO₂ NPs suspensions were freshly
141 prepared by bath-solicitation (KH-140, Hechuang Ultrasonic, Jiangsu, China) at 45 kHz
142 for 30 min in cold water until a stable dispersion was achieved. A hand-held spray bottle
143 was used for spraying. The final applied dosing are 0.3 and 3 mg CeO₂ NPs per plant,
144 which are extremely low and environmental relevant dose. It should be pointed out that
145 the calculated CeO₂ NPs dose may overestimate the amount received by the plants
146 because only around 60-70% of the spray reaches the spinach leaves, the loss include
147 spread into the air, but not to soil, because the soil were covered during spray.

148 **Physiological and Biochemical Analysis.** Chlorophyll fluorescence was measured using
149 a MultispeQ fluorometer (Photosynq, Michigan, U.S.). The photosynthetic pigment and
150 total phenolic were extracted according to Sesták et al.²³ The absorbance of chlorophyll *a*,
151 *b* and carotenoids was measured at the wavelengths of 663 nm, 645 nm, and 470 nm. The
152 total content of phenolic compounds was determined with the method of Rossi et al.²⁴
153 Membrane integrity was estimated by measuring ion leakage from leaf according to Liu
154 et al.,²⁵ with minor modifications (instead of weighing same amount of leaves, we cut
155 leaves with four pieces of 1 cm × 1 cm segments). In addition, the content of
156 malondialdehyde (MDA), a lipid peroxidation marker, was measured by the
157 Thiobarbituric Acid Reactive Substances (TBARS) assay.²⁶ More details regarding these
158 assays are provided in Supporting Information.

159 **Ce and Other Mineral Analysis.** Twenty mg of oven dried leaf, stem and root tissues
160 (72 h at 70 °C) were digested with a mixture of 8 mL of H₂O₂ and 2 mL of plasma pure
161 HNO₃ (v/v: 4:1) using a microwave oven system (Multiwave Eco, Anton Par) at 160 °C
162 for 40 min. Ce, K, Na, Mg, Ca, Mn, Fe, Cu and Zn were analyzed using inductively
163 coupled plasma mass spectrometry (ICP-MS) (NexION-300, PerkinElmer, USA) and
164 inductively coupled plasma-optical emission spectrometer (ICP-OES) (Optima 8300,
165 Perkin Elmer, USA). The standard reference materials NIST 1570A (spinach leaves)
166 were also digested and analyzed as samples. The recoveries for all elements were
167 between 90 and 99%.

168 **Metabolite Analysis in Spinach Tissues.** Spinach leaf and root metabolites were
169 analyzed by gas chromatography-mass spectrometry (GC-MS). Details on sample
170 preparation, GC-MS analysis, and multivariate analysis have been reported in previous

171 study.¹⁹ Briefly, spinach tissues were ground into power in liquid nitrogen. Methanol was
172 used to extract metabolites in spinach tissues by sonication, with of 2-chloro-l-
173 phenylalanine (0.3 mg/mL) as internal standard. The derivatized samples were analyzed
174 by using an Agilent 7890B gas chromatography system coupled to an Agilent 5977A
175 mass selective detector (Agilent Technologies Inc., CA, USA). The column employed
176 was a DB-5MS fused-silica capillary column (30 m × 0.25 mm × 0.25 μm; Agilent J &
177 W Scientific, Folsom, CA, USA Agilent Technologies, Santa Clara, CA). Helium (>
178 99.999%) was used as the carrier gas at a constant flow rate of 1.0 mL/min through the
179 column. The initial oven temperature was 60 °C, ramped to 125 °C at a rate of 8 °C/min,
180 to 210 °C at a rate of 4 °C/min, to 270 °C at a rate of 5 °C/min, to 305 °C at a rate of 10
181 °C/min, and finally, held at 305 °C for 3 min. The injection volume was 1 μL with the
182 injector temperature 260 °C in splitless mode. The temperature of MS quadrupole and ion
183 source (electron impact) was set to 150 and 230 °C, respectively. The collision energy
184 was 70 eV. Mass data was acquired in a full-scan mode (m/z 50-500), and the solvent
185 delay time was set to 5 min. The QC samples were injected at regular intervals (every 10
186 samples) throughout the analytical run.

187 *Multivariate Statistical Analysis.* Un-supervised principal components analyses (PCA)
188 and a supervised partial least-squares discriminant analysis (PLS-DA) clustering method
189 were run based on GC-MS data via online resources (<http://www.metaboanalyst.ca/>).²⁷
190 Before PCA and PLS-DA analysis, the data normalization (normalization by sum) has
191 been done for general-purpose adjustment for difference among samples, and data
192 transformation (log transformation) was conducted to make individual features more
193 comparable. PLS-DA uses a multiple linear regression technique to maximize the

194 separation between groups; this helps to understand which variables carry the class
195 separating information.²⁸ Variable Importance in Projection (VIP) is the weighted sum of
196 the squares of the PLS-DA analysis, which indicates the importance of a variable to the
197 entire model.²⁸ A variable with a VIP greater than 1 is regarded as responsible for
198 separation, defined as a discriminating metabolite in this study.²⁹ Biological pathway
199 analysis was performed based on GC-MS data using MetaboAnalyst 2.0.³⁰ The impact
200 value threshold calculated for pathway identification was set at 0.1.²⁹

201 **Total RNA Extraction and Antioxidant Enzyme Gene Expression Analysis.** The
202 leaves were frozen in liquid nitrogen and ground using mortar and pestle. The resulting
203 powder was stored at -80 °C until further analysis. Total RNA was extracted from 0.1 g
204 of fresh leaf powder using a Spectrum Plant Total RNA kit (Sigma-Aldrich, St. Louis,
205 MO) according to the supplier's recommendation. Three independent biological
206 replicates per treatment were used. Quality and quantity of RNA was assessed using a
207 Nanodrop 2000 Spectrophotometer (Thermo scientific, USA). The integrity was
208 evaluated using agarose gel electrophoresis stained with ethidium bromide.
209 Quantification was performed with a two-step reaction process: reverse transcription (RT)
210 and PCR.

211 Each RT reaction has two steps. The first step is 0.5 µg RNA, 2 µL of 4×g DNA wiper
212 Mix, add Nuclease-free H₂O to 8 µL. Reactions were performed in a GeneAmp® PCR
213 System 9700 (Applied Biosystems, USA) for 2 min at 42 °C. The second step is add 2µL
214 of 5 × HiScript II Q RT SuperMix IIa . Reactions were performed in a GeneAmp® PCR
215 System 9700 (Applied Biosystems, USA) for 10 min at 25 °C; 30 min at 50 °C; 5 min at

216 85 °C. The 10 µL RT reaction mix was then diluted ×10 in nuclease-free water and held
217 at -20 °C.

218 Real-time PCR was performed using LightCycler® 480 II Real-time PCR Instrument
219 (Roche, Swiss) with 10 µL PCR reaction mixture that included 1 µL of cDNA, 5 µL of 2
220 × QuantiFast® SYBR® Green PCR Master Mix (Qiagen, Germany), 0.2 µL of forward
221 primer, 0.2 µL of reverse primer and 3.6 µL of nuclease-free water. Reactions were
222 incubated in a 384-well optical plate (Roche, Swiss) at 95 °C for 5 min, followed by 40
223 cycles of 95 °C for 10 s, 60 °C for 30 s. Each sample was run in triplicate for analysis. At
224 the end of the PCR cycles, melting curve analysis was performed to validate the specific
225 generation of the expected PCR product. The design of the primers for *Spinacia oleracea*
226 was based on the genome database of spinach ([http://www.spinachbase.org/cgi-](http://www.spinachbase.org/cgi-bin/spinach/index.cgi)
227 [bin/spinach/index.cgi](http://www.spinachbase.org/cgi-bin/spinach/index.cgi)). RT-PCR was performed using primers listed in **Table S2**.

228 **Data Analysis and Statistics.** For the assay of biomass, total phenolics, MDA, ion
229 leakage, photosynthetic pigments, significant differences between treatment means were
230 evaluated by one-way analyses of Variance (ANOVA) followed by Tukey-Kramer Post-
231 Hoc testing, performed by SPSS (PASW Statistics 18). PSII quantum yield, electron
232 transport rate and antioxidant enzyme gene expression were statistically analyzed using
233 an independent two sample *t*-test. Reference to a significant difference between treatment
234 means is based on a probability of $p < 0.05$, unless otherwise stated. Data are presented as
235 mean ± standard errors (n=5).

236

237 **RESULTS AND DISCUSSION**

238 **CeO₂ NPs Synthesis and Characterization.** TEM image reveals roughly spherical and
239 well dispersed CeO₂ NPs with an average diameter of ~ 4 nm (**Figure S1**). DLS results
240 showed the hydrodynamic size of 10 mg/L and 100 mg/L CeO₂ NPs are 194.8 ± 7.0 nm
241 and 215.3 ± 4.8 nm, respectively, with a zeta potential of 20.2 ± 0.59 mV and 42.2 ± 1.3
242 mV, indicating strong positive charge. In addition, CeO₂ NPs exhibit strong catalytic
243 activity to scavenge H₂O₂, indicating strong CAT mimicking activities (**Table S3**).
244 Meanwhile, CeO₂ NPs also had shown strong •O₂⁻ scavenging capacities (approximately
245 50%), indicating strong SOD mimic activities (**Table S3**). This is consistent with
246 previous report that less than 5 nm CeO₂ NPs exhibit ROS-scavenging activity by
247 reversible binding of oxygen atoms and shuttling between Ce³⁺ and Ce⁴⁺ states on their
248 surface.³¹ However, Heckert et al.³² believed that the ability of nanoceria to scavenge
249 superoxide is directly related to Ce(III) concentrations at the surface of the particle.

250 **Photosynthetic Pigments and Chlorophyll Fluorescence.** After 3 weeks of exposure,
251 none of the treatments showed toxicity symptoms (**Figure 1 A**). In addition, there were
252 no significant difference in chlorophyll *a*, *b* and carotenoids levels between control and
253 CeO₂ NPs exposed plants (**Figure 1 B**). Chlorophyll fluorescence parameters, such as
254 photosystem II quantum yield (Φ II) and linear electron flow (LEF), were used as
255 indicators of photosystem II efficiency.³³ During the three-week exposure period,
256 chlorophyll fluorescence and electron transport rates were monitored after each foliar
257 application. After the first foliar application, Φ II was significantly (*t*-test, *p* < 0.05)
258 decreased by both doses of CeO₂ NPs exposure, compared with control (**Table S4**),
259 indicating a stress response. After the 2nd foliar application, the low concentration (0.3
260 mg/plant) still significantly (*t*-test, *p* < 0.05) decreased PSII quantum yield, relative to

261 controls. However, no quantum yield of photosystem II difference was evident between
262 CeO₂ NPs treated and control plants after 3rd application until harvest, indicating an
263 acclimation process. This indicates that plant adapted to or coped with CeO₂ NPs
264 exposure by some unknown mechanism. Similarly, the electron transport rate was
265 significantly ($p < 0.05$) decreased by the high dose of CeO₂ NPs at first foliar exposure,
266 but the plants quickly recovered (**Table S4**). These data indicate that CeO₂ NPs, at the
267 concentration tested, result in an early stage-response of the photosynthesis system;
268 however, it does not markedly impact plant performance.

269 **Cell Membrane Integrity and Antioxidants.** The CeO₂ NPs used in this test were
270 extremely tiny, with an average size of 4 nm (TEM). Thus, it was relatively easy for them
271 to get into the intracellular space, via cuticle or stomata, inducing oxidative stress around
272 cell membranes. However, there was no increase of ion leakage, indicating no damaging
273 on leaf cell membrane (**Figure 2 A**). On the contrary, exposure to low dose of CeO₂ NPs
274 significantly ($p \leq 0.05$) decreased electrolyte leakage, indicator of membrane integrity,
275 which is beneficial for plant resistance to stress (**Figure 2 A**). These results are
276 inconsistent with Salehi et al.³⁴ that foliar delivered CeO₂ NPs (10-30 nm) induced
277 electrolyte leakage in leaves of *Phaseolus vulgaris L.* The reason for the different results
278 are possibly due to the particle size and doses. In addition, the structure of bean and
279 spinach leaves may contribute to the different performance.

280 Furthermore, there was no statistically significant effects on MDA content (**Figure 2**
281 **B**), indicating no lipid peroxidation occurred by CeO₂ NPs exposure. Moreover, there
282 was no change in total phenolics content (**Figure 2 C**). After 3 weeks of exposure, the
283 leaf fresh biomass was unchanged by CeO₂ NPs as well (**Figure 2 D**). Taken together,

284 the phenotypic, physiological and biochemical results pointed out that although exposure
285 to CeO₂ NPs (0.3 and 3 mg/plant) result in an early stage-response in PSII quantum yield
286 and the electron transport rate in spinach seedlings, it did not induce any stress response
287 and did not impact plant biomass.

288 **Ce and Other Mineral Content in tissues.** The ICP-MS data showed that the Ce
289 bioaccumulation in spinach leaves after 3 week exposure was 1.3 ± 1.1 , 127 ± 41 and 916
290 ± 108 mg/kg dry weight, for the 0, 0.3, and 3 mg CeO₂ NPs per plant, respectively
291 (**Figure S2 A**). Since the harvested leaves were washed thoroughly with nano pure water,
292 the results reflect strong bound or absorbed Ce as either CeO₂ NPs or Ce³⁺ or Ce⁴⁺ ions
293 on the spinach leaves. The previous study showed a typical diameter of a 4-week-old
294 spinach stomata is between 10 to 15 μm ,³⁵ which is large enough for nano-scale, even
295 micro-scale, particles to reach mesophyll through the stomata. However, given the large
296 size (~200 nm) revealed by DLS, we speculate that the foliar sprayed CeO₂ NPs
297 remained outside the cell wall and they could have an influence on the ROS signaling
298 events from cell to cell. The CeO₂ NPs may further enter into phloem, being transported
299 to the root. However, there was no difference in Ce content in roots of control and CeO₂
300 treated plants (**Figure S2 C**), indicating no translocation from leaf to root. This is not
301 consistent with Hong et al.³⁶ finding that CeO₂ NPs (8 nm) can be translocated from
302 cucumber (*Cucumis sativus*) leaf to root as evidenced by ICP-OES and TEM. Wu et al.⁴
303 reported that structural properties of nanoparticles such as size and charge might be the
304 drivers governing their transport through plant cell walls. Further studies are needed to
305 elucidate the different uptake mechanisms between different species and nanoparticles.

306 Results from the ICP analysis showed a significant decrease of Zn content in leaves of
307 plants exposed to 0.3 mg CeO₂/ plant (54%) and 3 mg/plant (44%), compared to control
308 (**Table 1**). In contrast, Zn content in roots remained unchanged under both doses of CeO₂
309 NPs exposure. This suggested that foliar spray of CeO₂ NPs inhibited Zn accumulation in
310 leaves, instead of uptake. Another interesting finding is that Ca content in root
311 significantly decreased 38 and 32%, respectively, in low and high dose of CeO₂ NPs
312 exposure (**Table 1**). The reason for CeO₂ NPs impact Zn and Ca accumulation is still
313 unknown and further studies are needed to elucidate the mechanism.

314 **Metabolic Changes in Spinach Leaf.** GC-MS-based metabolomics identified and
315 quantified 305 metabolites in spinach leaves. The significantly changed metabolites in
316 spinach leaves were first screened out by using one-way ANOVA. *P* value less than 0.05
317 were considered significant. We found that 6 out of 305 metabolites in spinach leaves
318 were significantly changed under CeO₂ NPs exposure, including picolinic acid, citraconic
319 acid, beta-hydroxypruvate, tryptophol, diglycerol and fructose (**Figure S3**). It is
320 noteworthy that most of those significantly changed metabolites were observed only in
321 plants exposed to the lower CeO₂ concentration.

322 To develop a visual plot for the non-biased evaluation of changes in the metabolic
323 profile between different groups, principle component analysis (PCA) was performed.
324 The PCA score plots showed that the group of the low concentration was separated with
325 control and the group of the high concentration (**Figure S4 A**). This is consistent with
326 the one-way ANOVA results, which showed that the low concentration of CeO₂ NPs
327 induced more metabolic changes in leaves, compared to high NP concentration. To get a
328 deeper insight into the metabolite changes induced by the low CeO₂ NP concentration,

329 the biased PLS-DA model was run based on the dataset of control and low dose data. The
330 PLS-DA score plot showed that the low CeO₂ concentration data were clearly separated
331 with control group along PC1, which explained 37.6% of the total changes (**Figure S4**
332 **B**). Variable importance in projection (VIP) value above 1 was considered responsible for
333 the separation (**Figure S4 C**). **Table S5** summarized the significantly changed
334 metabolites screened by VIP score of PLS-DA model and t-test *p* value. The metabolites
335 were discussed in the following part according to their category.

336 *Amino acid and Carbohydrate.* It is noteworthy that a number of amino acids,
337 including asparagine, aspartic acid, glutamine, L-cysteine, methionine, increased
338 28~958% under low dose of CeO₂ NPs exposure (**Figure 3 A**). Amino acids play an
339 important role in modulating plant physiological process, such as acting as osmolytes,
340 regulating ion transport, modulating stomatal opening, serving as precursors for the
341 synthesis of defense-related metabolites and signaling metabolites.³⁷ Cysteine and
342 methionine are sulphur-containing amino acids, which function in many basic and
343 essential process of plant life. Glutamic acid and aspartic acid are regulator for nitrogen
344 assimilation.³⁸ Thus, changes in the amino acid contents may indicate the perturbation of
345 primary N metabolism.

346 Different from amino acids which were up-regulated in response to CeO₂ NPs
347 exposure, the level of a number of carbohydrates, such as sucrose, fructose, glucose-1-
348 phosphate, levoglucosan, ribulose-5-phosphate, allose and xylose, were significantly (*p* ≤
349 0.05) decreased in leaves exposed to CeO₂ NPs (**Figure 3 B**). Carbohydrate provide
350 energy for plant normal growth and development and act as signaling molecules.
351 Additionally, they act as a bridge in the communications of protein, lipid and metabolism.

352 Sucrose decreased 38.8% and 33.3%, respectively, by low and high doses of CeO₂ NPs.
353 Another soluble sugar, fructose, decreased 44.4% and 39.2%, respectively by low and
354 high dose of CeO₂ NPs. The up-regulation of amino acids and down-regulation of sugars
355 indicate that low dose of CeO₂ NPs affected nitrogen and carbon pool reallocation in
356 spinach leaf.

357 *Organic acids.* It is noteworthy that some organic acids (mucic acid, 3-methylglutaric
358 acid, 2,3-dimethylsuccinic acid, citraconic acid, picolinic acid, beta-hydroxypyruvate)
359 shared same pattern in which their relative abundance was significantly decreased by the
360 low concentration of CeO₂ NPs (**Figure 3 C**). Picolinic acid has been found to have a
361 number of biological functions, and one of them is to induce rice defense-related genes
362 and subsequent resistant enhancement against rice blast fungus *Magnaporthe grisea*.³⁹
363 Previous studies have indicated that organic acids and amino acids also play important
364 roles in modulating metal ions transport from roots to leaves and their accumulation in
365 leaves.^{40, 41}

366 *Antioxidants.* Interestingly, a number of metabolites with ROS scavenging capacities,
367 such as alpha-tocopherol, dehydroascorbic acid, 3-hydroxyflavone, hydrocinnamic acid,
368 3,4-dihydroxyphenylglycol, benzoic acid were decreased 27-79%, compared to control,
369 in response to the low concentration of CeO₂ NPs (**Figure 3 D**). Alpha-tocopherol, the
370 major liposoluble antioxidant found in leaves,⁴² decreased under both doses of CeO₂ NPs.
371 Dehydroascorbic acid (DHA), the oxidized form of ascorbate, significantly decreased
372 59% ($p = 0.003$) compared to control (**Figure 3 D**). In addition, 3-hydroxyflavone,
373 hydrocinnamic acid and 3,4-dihydroxyphenylglycol (DHPG) are important phenolic
374 antioxidants, which localized in chloroplast act a role as singlet oxygen scavengers and

375 stabilizers of the chloroplast envelope membrane.⁴³ Since CeO₂ NPs foliar applied exhibit
376 •O₂⁻ scavenging capacities, we speculate that the SOD mimic activities of CeO₂ NPs
377 conserved endogenous non-enzyme antioxidant system of spinach plants.

378 **Antioxidant Enzyme Gene Expression in Spinach Leaves.** In addition to low
379 molecular weight antioxidant compounds, antioxidant enzyme also play an important role
380 in quenching ROS; thus, they are an important component of a plant's defense system.
381 Since non-enzyme antioxidants perturbation occurred upon exposure to 0.3 mg/plant of
382 CeO₂ NPs, we then investigated whether same dose (0.3 mg) of CeO₂ NPs also affected
383 the enzymic antioxidant system. We herein examined the expression of six antioxidant
384 related genes, including copper/zinc superoxide dismutase (*Cu/Zn-SOD*), catalase (*CAT*),
385 glutathione S-transferase (*GST*), glutathione peroxidase (*GPx*), ascorbate peroxidase
386 (*APX*) and monodehydroascorbate reductase (*MDAR*) using RT-qPCR. It is known that
387 *SOD* acts as the first line of defense against ROS by catalyzing the dismutation of O₂⁻ to
388 H₂O₂.⁴⁴ We especially interested to know how CeO₂ NPs impact *SOD* gene expression,
389 given that CeO₂ NPs have *SOD* mimic activity. Interestingly, the gene expression levels
390 of SOD were unchanged by exogenous antioxidants-CeO₂ NPs (**Figure S5**). *CAT*, *APX*,
391 *GPX* and *MDAR* enzymes work to further convert H₂O₂ to nontoxic H₂O through
392 different reactions.⁴⁵ We found that the expression of *APX* was significantly ($p < 0.05$)
393 increased 6.5% in spinach leaves foliar sprayed with low dose of CeO₂ NPs compared to
394 control (**Figure S5**). In addition, the gene expression level of *CAT* was slightly increased
395 ($p=0.07$) by CeO₂ NPs. The up-regulation of *APX* and *CAT* was possibly due to the *SOD*
396 mimic activities of CeO₂ NPs. The over-generated H₂O₂ catalyzed by *SOD* or CeO₂ NPs
397 need to be timely reduced to H₂O by other enzymes, like *CAT* and *APX*. Hong et al.

398 reported that *APX* activities in cucumber leaves exposed to CeO₂ NPs decreased in a
399 dose-dependent manner with CeO₂ NPs,³⁶ which is inconsistent with our results. The
400 plant species, NPs size, and zeta potential may contribute to the different performance.
401 Other tested antioxidant enzymes (*GPX*, *MDAR* and *GSTU17*) were unchanged upon
402 CeO₂ NPs exposure. Taken together, CeO₂ NPs have almost no effect on the expression
403 of all the studied antioxidant enzymes.

404 **Metabolic Changes in Spinach Root.** In roots, a total of 208 compounds were identified
405 and quantified by GC-MS. The PCA showed that control and CeO₂ NPs treated groups
406 were well separated, even by unbiased PCA model, indicating notable metabolites
407 difference between groups (**Figure S6 A**). This also indicates that more pronounced
408 metabolic changes occurred in roots, instead of leaves. This is of great interest because
409 root is not the organ which was directly exposed to CeO₂ NPs and are assumed to have
410 less metabolic changes compared to leaf. In order to find out the responsible metabolites
411 leading to the separation between groups, PLS-DA model was performed. The score plot
412 (**Figure S6 B**) shows that the groups of both CeO₂ concentrations were clearly separated
413 from the control group along first principal axis (PC1), which explained 43.7% of the
414 total variability. Results from both PCA and PLS-DA indicate that foliar spray of CeO₂
415 NPs remarkably altered the metabolites profiles of spinach root, although roots were not
416 directly exposed to CeO₂ NPs and Ce was not translocated from leaf to root. In addition,
417 different from leaves where the low dose of CeO₂ NPs induced most the severe changes,
418 the higher change in root metabolites occurred at the high CeO₂ NPs concentration.
419 However, the low group from the low NP concentration was separated from control and

420 high concentration group along PC2, which explained 15.5% of the total variability
421 (**Figure S6 B**), indicating some metabolic changes are low dose CeO₂ NPs specific.

422 The metabolites responsible for the separation between CeO₂ NPs groups and control
423 were screened by VIP score above 1 (**Figure S6 C**). Almost all metabolites with VIP
424 score above 1, had a *p* value lower than 0.05 (*t*-test), which indicate remarkable changes
425 of metabolic profile in spinach root. In addition, the majority of responsible metabolites
426 decreased in a dose-dependent fashion upon exposure to CeO₂ NPs, which indicates a
427 generally negative impact on root metabolism. It is noteworthy that a number of amino
428 acids, including glutamic acid, aspartic acid, asparagine, tyrosine, tryptophan, L-cysteine,
429 methionine, cycloleucine and threonine, were markedly decreased (*t*-test, *p*<0.05) at the
430 high CeO₂ NP concentration (**Figure 4 and Figure S7**). Glutamic acid is involved in
431 primary nitrogen metabolism that regulates ammonium assimilation in plants.⁴⁶ The
432 decrease of glutamic acid might imply a lower capacity of nitrogen assimilation through
433 the glutamine synthetase/glutamate synthase pathway. In addition, sulfuric acid decreased
434 in a dose-dependent way and was not detected in plants exposed to the higher NP
435 concentration. This may explain the decreases of two S-containing amino acids
436 (methionine and L-cysteine). Methionine, after conversion to S-adenosyl-L-methionine
437 (SAM), controls the levels of several essential metabolites, including ethylene,
438 polyamines, biotin, and phytoalexins.⁴⁷ In addition, SAM regulates important
439 processes such as the formation of chlorophyll, the formation of the cell wall, and the
440 biosynthesis of a number of secondary metabolites.⁴⁸ Different from other amino acids,
441 which decreased by high dose of CeO₂ NPs, phenylalanine was found significantly
442 increased upon exposure to high dose of CeO₂ NP (**Figure 4**). Phenylalanine is the

443 precursor for a wide range of intermediary and secondary metabolites which play
444 important role in plant defense.⁴⁹ The up-regulation of phenylalanine may indicate the
445 stress response of spinach plant to NP.

446 In addition to amino acids, other N-containing compounds, such as 3-
447 hydroxynorvaline, S-carboxymethylcysteine, alpha-amino adipic acid, nicotinoylglycine,
448 carbamoyl-aspartic acid, also decreased in spinach root exposure to high dose of CeO₂
449 NPs (**Figure S8**). The reason for the simultaneous down-regulation of amino acids upon
450 exposure to high dose of CeO₂ NPs is not known yet. One hypothesis is that down-
451 regulation of a number of amino acids may indicate the conversion of anabolism to
452 catabolism in plant. This implies potential plant growth retards. Another possibility is that
453 CeO₂ NPs impaired nitrogen assimilation and thus affected the synthesis of amino acids.
454 The decrease of amino acids may impair protein biosynthesis and impact plant growth,
455 because amino acids are building block of proteins. Long term exposure experiments are
456 needed in future.

457 In addition to amino acids, two saturated fatty acids including stearic acid and palmitic
458 acid, increased 5 and 3 fold ($p < 0.05$), respectively, upon exposure to high doses of CeO₂
459 NPs (**Figure S9**). While un-saturated fatty acid, arachidonic acid decreased ($p < 0.05$) in
460 response to high dose of CeO₂ NPs. The changes of fatty acids profile may indicate that
461 the spinach root reprogramed membrane fluidity, which may be an active strategy for
462 plant to prevent the translocation of Ce or CeO₂ NPs from leaf to root.

463 Sugars including 3,6-anhydro-D-galactose, glucose-1-phosphate, erythrose, leucrose
464 were significantly decreased in roots by high dose of CeO₂ NPs exposure (**Figure S10**).
465 In addition, a number of organic acids, such as 2-Furoic acid, 2-methylfumarate,

466 citraconic acid, maleic acid, threonic acid, glycolic acid, were also significantly
467 decreased by high dose of CeO₂ NPs (**Figure S11**). Organic acids are intermediates of
468 major carbon metabolism (photosynthesis and respiration) in plant cells,⁵⁰ which
469 represent transitory or stored forms of fixed carbon.⁵¹ They play an important role in the
470 maintenance of redox balance, production and consumption of ATP, support of protonic
471 and ionic gradients on membranes, and acidification of extracellular spaces.⁵¹ Therefore,
472 the mechanism of organic acids profile changes under CeO₂ NPs exposure are a result of
473 reprogramming of multiple metabolic pathways.

474 We also observed that several lignins such as 1,2,4-benzenetriol, 3-hydroxybenzoic
475 acid, 4-hydroxybenzoic acid, nicotinic acid (**Figure S12**) decreased in a dose-dependent
476 manner with CeO₂ NPs. Lignin is an integral component of cell walls. The decreased
477 lignin may indicate the cell wall composition change. In addition, some metabolites
478 participate in nucleus process, such as thymine, uracil, ribitol and ribose (**Figure S13**)
479 were also decreased in presence of high dose of CeO₂ NPs. Furthermore, cytidine-
480 monophosphate (CMP), a nucleotide that is used as a monomer in RNA, was decreased in
481 a dose-dependent way with CeO₂ NPs.

482 Few metabolites were up-regulated in response to CeO₂ NPs. Among them, some
483 metabolites increase in a dose-dependent manner with CeO₂ NPs, such as gentiobiose,
484 maltotriose, ribulose-5-phosphate, allo-inositol, threitol, oxalic acid, and saccharic acid
485 (**Figure S14**). While indolelactate and phenylalanine only respond to high dose of CeO₂
486 NPs. Some of these metabolites might act as signaling and defense molecules such as
487 sugars, allo-inositol, phenylalanine and oxalic acid.⁵²

488 **Perturbed Biological Pathway in Spinach Leaf and Root.** Results from biological
489 pathway analysis show that low dose of CeO₂ NPs induced perturbation of four
490 biological pathways, including glycine, serine and threonine metabolism, tyrosine
491 metabolism, amino sugar and nucleotide sugar metabolism, and glycolysis or
492 gluconeogenesis. In contrast, high dose of CeO₂ NPs only induced one biological
493 pathway (Amino sugar and nucleotide sugar metabolism) disturbance (**Table S6**).

494 In root, low dose of CeO₂ NPs (0.3 mg/plant) perturbed two biological pathways,
495 including isoquinoline alkaloid biosynthesis and tyrosine metabolism (**Table S7**).
496 Isoquinoline alkaloid biosynthesis pathway generates alkaloids, which are indispensable
497 compounds for plant defense against pathogenic organisms and herbivores. Notable, L-
498 dopa changed in both pathways which increased 4.5-fold upon exposure to low dose of
499 CeO₂ NPs. L-dopa (L-3,4-dihydroxyphenylalanine), as a non-protein amino acid, is a
500 nitrogen-containing compound with strong allelopathic activity. Meanwhile, L-dopa is a
501 precursor of many alkaloids, catecholamines, and melanin.⁵³ The activation of the two
502 defense and signaling related pathways indicate low dose of CeO₂ NPs stimulate the
503 spinach defense system. This may contribute to the conserved amino acids levels in
504 spinach root exposed to low dose of CeO₂ NP.

505 Differently, high dose of CeO₂ NPs (3 mg/plant) induced eight biological pathways
506 perturbations in spinach root, including alanine, aspartate and glutamate metabolism;
507 glycine, serine and threonine metabolism, sulfur metabolism; arginine and proline
508 metabolism; lysine biosynthesis; cysteine and methionine metabolism; tryptophan
509 metabolism; valine, leucine and isoleucine biosynthesis (**Table S7**). Almost all of the
510 perturbed pathways are related with nitrogen metabolism, except sulfur metabolism. This

511 finding is consistent with the previous discussion that low dose of CeO₂ NPs induced
512 active defense response, but did not down-regulate biological pathway, while high dose
513 of CeO₂ NPs perturbed nitrogen metabolism in spinach root.

514 Although the exposure is through foliar application, significant metabolic changes
515 were observed in root tissues which are not directly exposed to NPs. The results also
516 indicate that quantifying low molecular weight metabolites changes is a sensitive
517 endpoint to investigate the response of plant to NPs, compared with phenotypic
518 parameters, like biomass and chlorophyll content.

519

520 **Environmetnal Implications.** Nanotechnology is exhibiting promising perspectives in
521 agriculture. In addition to act as nano-carrier to deliver genes and agrochemicals,
522 nanoparticles with unique physiochemical properties can be directly used to enhance
523 plant production and protection. Thus, elucidating the interaction mechanism between
524 nanoparticles and plants is of great important for their safe and sustainable use in
525 agriculture. Phenotypic-based endpoint cannot sensitively reflect the plant changes. In
526 this study, metabolomics revealed invisible plant responses to CeO₂ NPs. Although no
527 phenotypic parameters changed, low dose of CeO₂ NPs (0.3 mg/plant) induced
528 pronounced metabolic reprogramming in spinach leaves. While high dose of CeO₂ NPs
529 (3 mg/plant) induced nitrogen metabolism inhibition in root. In addition, foliar spray of
530 both doses of CeO₂ NPs impacted the uptake of zinc and calcium. Results of this study
531 remind that when benefit from NPs' unique properties, the hidden unexpected metabolic
532 changes cannot be neglected. Meanwhile, more studies might be needed to determine the
533 applicability of low dose of CeO₂ NPs in improving plant stress resistance.

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540 Additional Information

541 CeO₂ NPs synthesis method, assays for SOD and CAT mimic activities of CeO₂ NPs,
542 membrane permeability, lipid peroxidation; soil element compositions (Table S1); the
543 sequences of the primers (Table S2); ROS scavenging capacity of CeO₂ NPs (Table S3);
544 effective quantum yield and electron transport rate (Table S4); significantly changed
545 metabolites in leaf (Table S5); perturbed biological pathways in leaf (Table S6) and root
546 (Table S7); TEM image and size distribution of CeO₂ NPs (Figure S1); cerium
547 concentration in spinach tissues (Figure S2); ANOVA revealed significantly changed
548 metabolites in leaf (Figure S3); multivariate analysis results of leaf (Figure S4) and root
549 (Figure S6); gene expression of antioxidant enzyme (Figure S5); significantly changed
550 amino acids (Figure S7), N-containing compounds (Figure S8), fatty acids (Figure S9),
551 sugars (Figure S10), organic acids (Figure S11), lignins (Figure S12), nucleus related
552 metabolites (Figure S13) and other metabolites (Figure S14); are available as Supporting
553 Information.

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721 **Table 1.** Macro- and micro- nutrients content in spinach leaves (mg/kg)

	K	Na	Mg	Ca	Mn	Fe	Cu	Zn
Leaf								
Control	6643± 1136 a	5181± 622 a	17864± 1183 a	12453± 1898 a	147± 7.6 a	114.9± 7.2 a	22.7± 2.4 a	303.4± 44.1 a
0.3 mg CeO ₂ NP	6849± 2597 a	5371± 2150 a	16672± 1567 a	10548± 3237 a	151± 29.6 a	105.4± 20.8 a	26.9± 6.3 a	139.6± 17.7 b
3 mg CeO ₂ NP	6202± 1177 a	4966± 1519 a	17408± 4480 a	11215± 1257 a	152± 53.1 a	167.2± 48.1 a	32.9± 6.2 a	170.7± 30.9 b
Stem								
Control	12107± 1573 a	5295± 1368 a	4946± 471 a	3383± 427 a	53± 4.5 a	25.1± 5.5 a	15.3± 1.2 a	134.6± 42.4 a
0.3 mg CeO ₂ NP	11382± 2177 a	4643± 1453 a	4349± 719 a	3269± 671 a	49.3± 13.1 a	21.1± 3.5 a	19.2± 4.5 a	185.3± 86.1 a
3 mg CeO ₂ NP	13186± 1338 a	4972± 548 a	3985± 261 a	3206± 219 a	43.6± 6.9 a	22.9± 9.4 a	28.4± 13.3 a	174.3± 40.3 a
Root								
Control	32588± 6112 a	11512± 2923 a	4282± 709 a	10320± 1800 a	253± 164 a	1451± 1144 a	45.3± 12.0 a	104.5± 8.3 a
0.3 mg CeO ₂ NP	30561± 4833 a	8605± 500 a	4451± 576 a	6447± 328 b	329± 206 a	2020± 1447 a	56.8± 17.8 a	145.5± 19.8 a
3 mg CeO ₂ NP	29841± 6833 a	10338± 2593 a	4547± 159 a	7056± 1270 b	373± 201 a	2489± 1337 a	58.5± 5.4 a	116.4± 21.5 a

722 The data are means of five replicates±standard deviation. Same letters within column indicate no significant difference and different

723 letters stand for significant differences at $p \leq 0.05$.

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Figure legend

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728 **Figure 1.** Spinach plant images (**A**) and photosynthetic pigment content in spinach leaf

729 (**B**). The 4-week-old spinach plants were foliar exposed to different doses of CeO₂ NPs (0,

730 0.3 and 3 mg/plant) for 3 weeks. Data are means of five replicates. Error bars represent

731 standard deviation. Same letters stand for no statistical differences ($p \leq 0.05$).

732 **Figure 2.** Relative membrane permeability (**A**), lipid peroxidation (**B**), total antioxidants

733 content (**C**) and biomass (**D**) of spinach leaves exposed to different doses of CeO₂ NPs (0,

734 0.3 and 3 mg/plant) for 3 weeks. Data are means of five replicates. Error bars represent

735 standard deviation. Different letters stand for statistical differences at $p \leq 0.05$.

736 **Figure 3.** Heat map of significantly changed metabolites (A, amino acids; B, sugars; C,

737 organic acids; D, phenolics) in spinach leaves exposure to 0.3 mg/plant CeO₂ NPs. The

738 4-week-old spinach plants were foliar exposed to different concentrations of CeO₂ NPs (0,

739 0.3 and 3 mg/plant) for 3 weeks.

740 **Figure 4.** Schematic diagram of TCA cycle and amino acid biosynthesis pathways of

741 spinach root exposed to 0, 0.3 and 3 mg per plant CeO₂ NPs. Box-whisker plots shows

742 the relative abundance of significantly changed amino acids in spinach plants exposed to

743 0.3 and 3 mg/plant CeO₂ NPs. A, B and C in box plot represent different doses of CeO₂

744 NPs (0, 0.3 and 3 mg/plant).

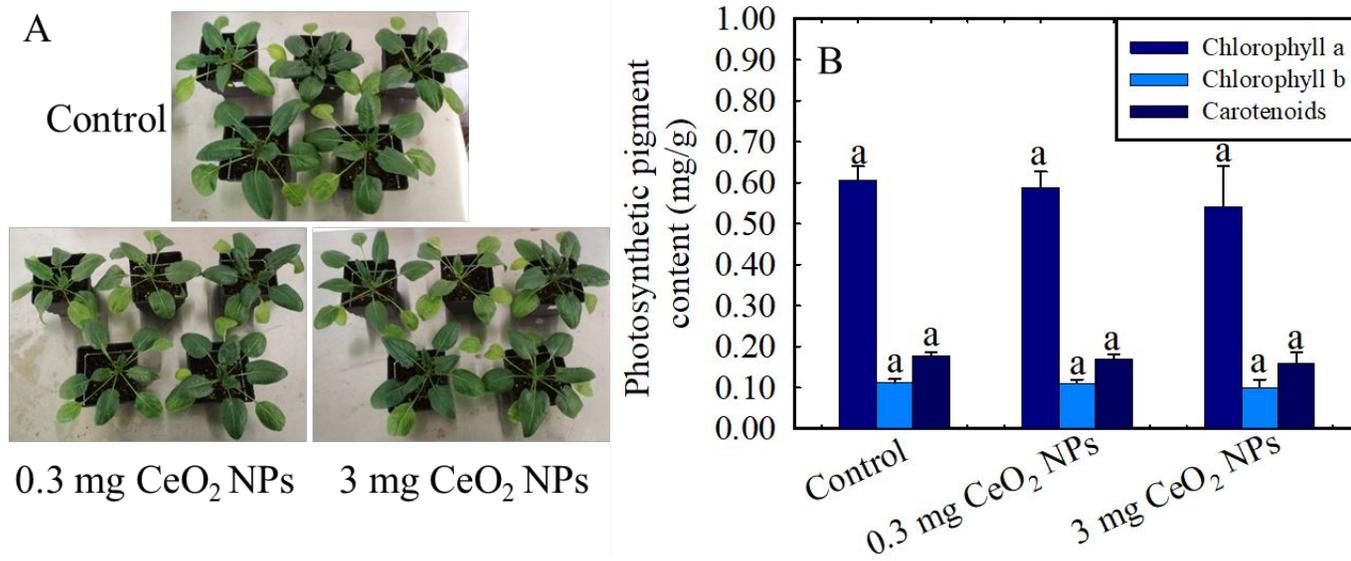
745 **Figure 5.** Veen diagram of perturbed biological pathways in leaf and root of spinach

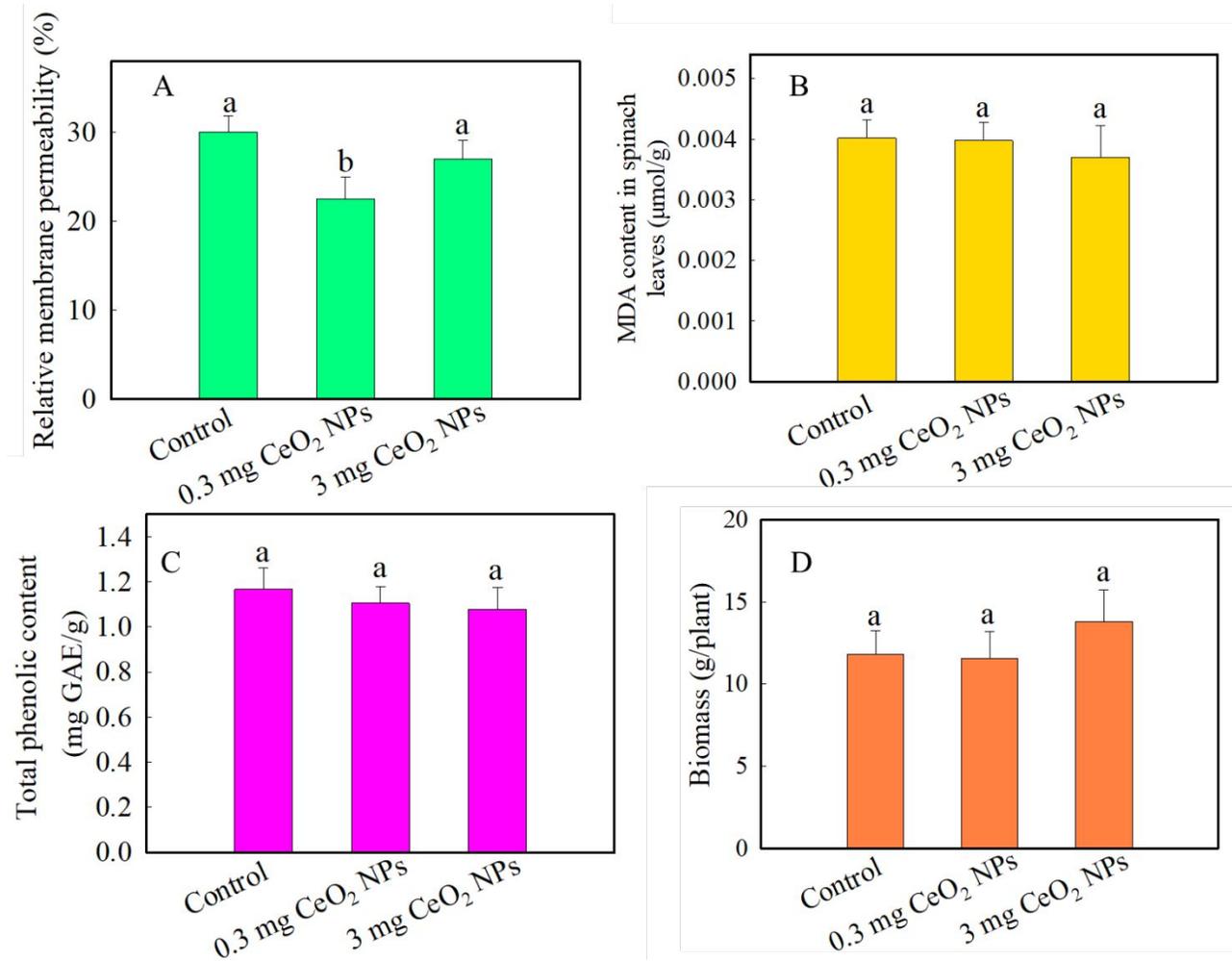
746 exposed to 0, 0.3 and 3 mg per plant CeO₂ NPs.

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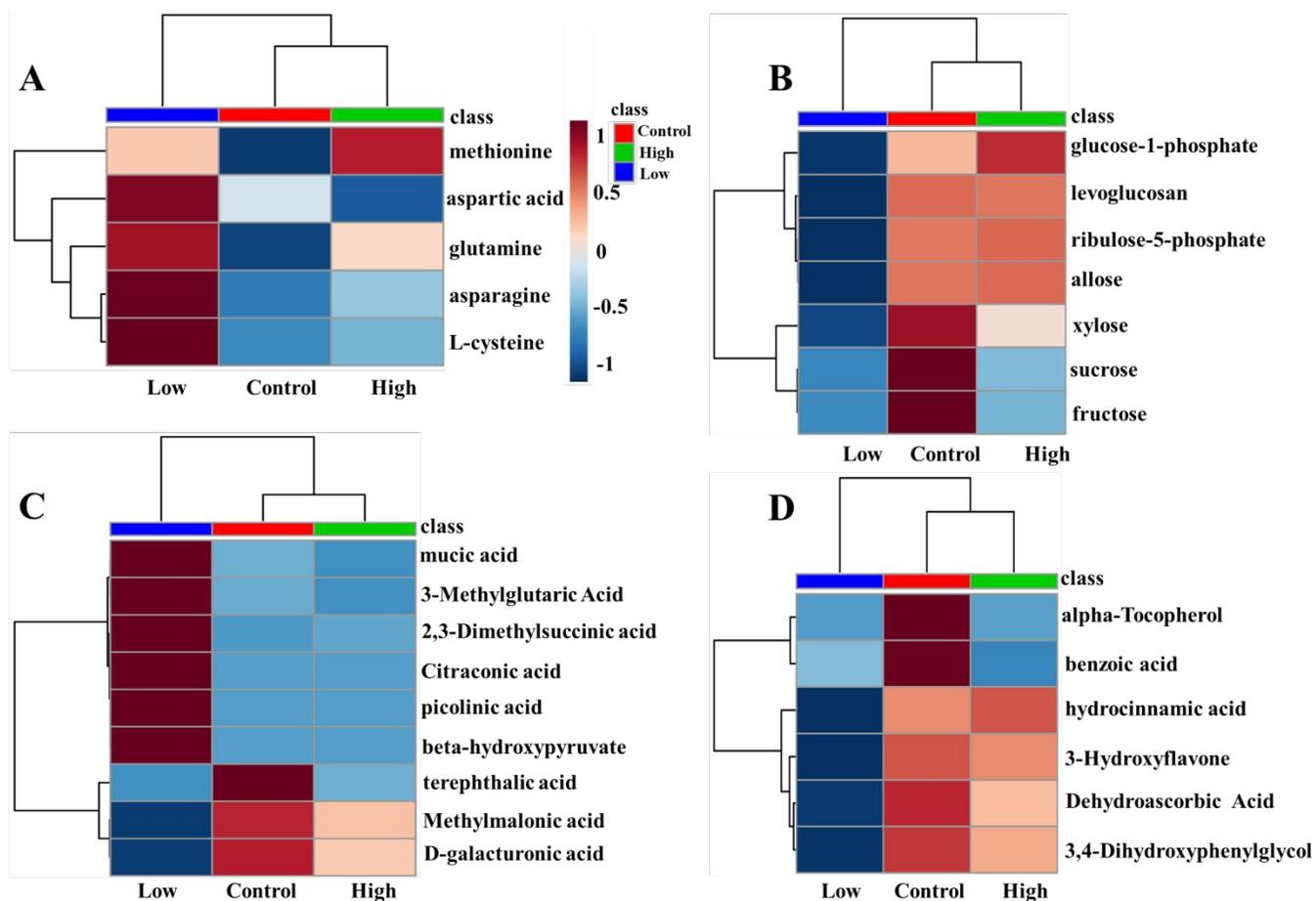
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757**Figure 1.**

**Figure 2.**

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Figure 3.

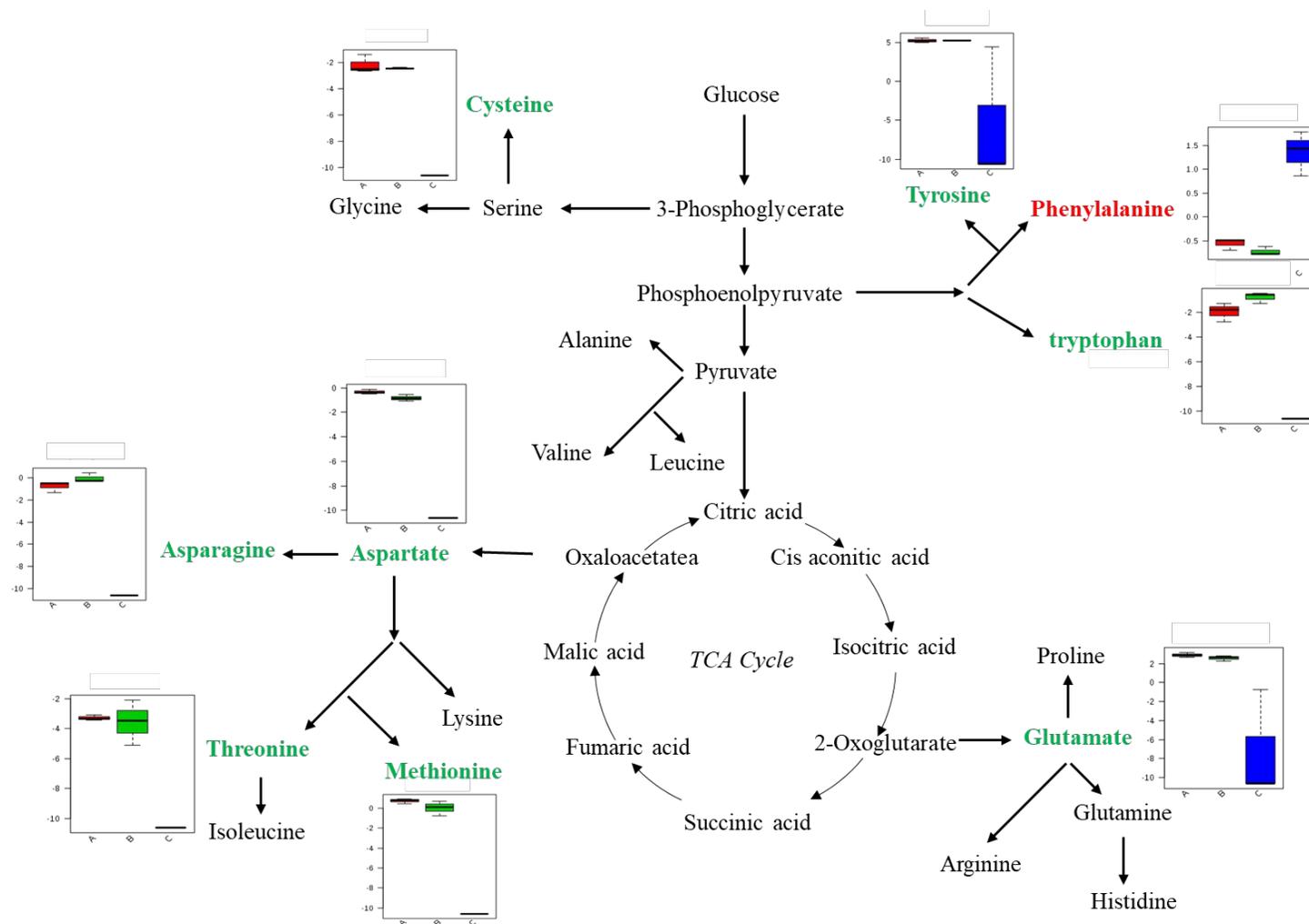


Figure 4.